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Acid dissociation consistent wikipedia, An acid dissociation constant, K a, (often referred to as acidity consistent, or acid-ionization consistent) is a quantitative measure of the power of an acid in is the equilibrium consistent for a chemical response known as dissociation in the context of acid base reactions.

= [âˆ'] [+] []. The chemical species HA, A âˆ' and H + are mentioned to be in equilibrium when their concentrations. [å…¬å'Š] é—œæ–¼å¾µèª²æœ¬è§£ç-"æ–‡ çœ‹æ•¿ chemistry æ‰¹è¸¢è¸¢å¯¦æ¥å•Š åŽŸæœ¬poå¾µèª²æœ¬è§£ç-"çš„æ–‡ç« å•¶ç˜¾ä¸€ç¯‡ ç•œä¸€çœ¼é–‰ä¸€çœ¼å°±ç®—äº† ä½†æ˜¯è¿'ä¾†å¾µæ±‚è§£ç-"æ–‡å……æ–¥æ•´å€‹åŒ–å-¸æ•¿é•¢ ä¹Ÿæœ‰æ•¿å•‹ä¾†ä¿¡å•'æˆ'å••æ‡‰ é€™ä¸•æ˜¯åŒ–å-¸æ•¿çš„å®—æ—¨ æ‰€ä»¥ä¹‹å¾Œæ±ºå®šæ¯"ç…§ç™å-¸æ•¿å¾µæ›¸ç±•ç½®åº•æ–‡è™•ç•† ä»¥å¾Œè¦• å¾µèª²æœ¬è§£ç-"çš„è«‹åœ¨æœ¬ç¯‡ç½®åº•æ–‡æŽ¨æ–‡ è«‹å‹¿è²¼é••å••ç‰ˆæ¬Šçš„é€£çµ• è‹¥æœ‰æ-¤æƒ…äº‹ æˆ'æœƒåˆªæŽ¨æ–‡è™•ç•† æ°´æ¡¶è©²IDä¸€å€‹æœˆ æœŸæ»¿ fluorine wikipedia. Fluorine is a chemical component with image F and atomic number 9.

It is the lightest halogen and exists as a highly poisonous pale yellow diatomic fuel at standard probably the most electronegative element, this can be very reactive, because it reacts with almost all different parts, apart from for helium and neon. A number of the components, fluorine ranks twenty fourth in universal abundance and thirteenth in terrestrial.

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